

07.06 Gauss Quadrature Rule for Integration-More Examples Civil Engineering

Example 1

The concentration of benzene at a critical location is given by

$$c = 1.75[\operatorname{erfc}(0.6560) + e^{32.73} \operatorname{erfc}(5.758)]$$

where

$$\operatorname{erfc}(x) = \int_{\infty}^x e^{-z^2} dz$$

So in the above formula

$$\operatorname{erfc}(0.6560) = \int_{\infty}^{0.6560} e^{-z^2} dz$$

Since e^{-z^2} decays rapidly as $z \rightarrow \infty$, we will approximate

$$\operatorname{erfc}(0.6560) = \int_5^{0.6560} e^{-z^2} dz$$

- Use two-point Gauss Quadrature Rule to approximate the value of $\operatorname{erfc}(0.6560)$.
- Find the absolute relative true error for part (a).

Solution

- First, change the limits of integration from $[5, 0.6560]$ to $[-1, 1]$ using

$$\int_a^b f(x) dx = \frac{b-a}{2} \int_{-1}^1 f\left(\frac{b-a}{2}x + \frac{b+a}{2}\right) dx$$

gives

$$\begin{aligned} \int_5^{0.6560} f(z) dz &= \frac{0.6560-5}{2} \int_{-1}^1 f\left(\frac{0.6560-5}{2}z + \frac{0.6560+5}{2}\right) dz \\ &= -2.1720 \int_{-1}^1 f(-2.1720z + 2.8280) dz \end{aligned}$$

Next, get weighting factors and function argument values for the two point rule,

$$c_1 = 1.0000$$

$$z_1 = -0.57735$$

$$c_2 = 1.0000$$

$$z_2 = 0.57735$$

Now we can use the Gauss Quadrature formula

$$\begin{aligned} & -2.1720 \int_{-1}^1 f(-2.1720z + 2.8280) dz \\ & \approx -2.1720 [c_1 f(-2.1720z_1 + 2.8280) + c_2 f(-2.1720z_2 + 2.8280)] \\ & \approx -2.1720 [f(-2.1720(-0.57735) + 2.8280) + f(-2.1720(0.57735) + 2.8280)] \\ & \approx -2.1720 [f(4.0820) + f(1.5740)] \\ & \approx -2.1720 [(5.8003 \times 10^{-8}) + (0.083955)] \\ & \approx -0.18235 \end{aligned}$$

since

$$\begin{aligned} f(4.0820) &= e^{-4.0820^2} \\ &= 5.8003 \times 10^{-8} \\ f(1.5740) &= e^{-1.5740^2} \\ &= 0.083955 \end{aligned}$$

b) The absolute relative true error, $|\epsilon_t|$, is (Exact value = -0.31333)

$$\begin{aligned} |\epsilon_t| &= \left| \frac{-0.31333 - (-0.18235)}{-0.31333} \right| \times 100\% \\ &= 41.801\% \end{aligned}$$

Example 2

The concentration of benzene at a critical location is given by

$$c = 1.75 [\operatorname{erfc}(0.6560) + e^{32.73} \operatorname{erfc}(5.758)]$$

where

$$\operatorname{erfc}(x) = \int_{\infty}^x e^{-z^2} dz$$

So in the above formula

$$\operatorname{erfc}(0.6560) = \int_{\infty}^{0.6560} e^{-z^2} dz$$

Since e^{-z^2} decays rapidly as $z \rightarrow \infty$, we will approximate

$$\operatorname{erfc}(0.6560) = \int_5^{0.6560} e^{-z^2} dz$$

- Use three-point Gauss Quadrature Rule to approximate the value of $\operatorname{erfc}(0.6560)$.
- Find the absolute relative true error for part (a).

Solution

a) First, change the limits of integration from $[5, 0.6560]$ to $[-1, 1]$ using

$$\int_a^b f(x)dx = \frac{b-a}{2} \int_{-1}^1 f\left(\frac{b-a}{2}x + \frac{b+a}{2}\right)dx$$

gives

$$\begin{aligned} \int_5^{0.6560} f(z)dz &= \frac{0.6560-5}{2} \int_{-1}^1 f\left(\frac{0.6560-5}{2}z + \frac{0.6560+5}{2}\right)dz \\ &= -2.1720 \int_{-1}^1 f(-2.1720z + 2.8280)dz \end{aligned}$$

The weighting factors and function argument values are

$$c_1 = 0.55556$$

$$z_1 = -0.77460$$

$$c_2 = 0.88889$$

$$z_2 = 0.0000$$

$$c_3 = 0.55556$$

$$z_3 = 0.77460$$

and the formula is

$$\begin{aligned} &-2.1720 \int_{-1}^1 f(-2.1720z + 2.8280)dz \\ &\approx -2.1720 [c_1 f(-2.1720z_1 + 2.8280) + c_2 f(-2.1720z_2 + 2.8280) + c_3 f(-2.1720z_3 + 2.8280)] \\ &\approx -2.1720 \left[\begin{array}{l} 0.55556 f(-2.1720(-0.77460) + 2.8280) \\ + 0.88889 f(-2.1720(0.0000) + 2.8280) \\ + 0.55556 f(-2.1720(0.77460) + 2.8280) \end{array} \right] \\ &\approx -2.1720 [0.55556 f(4.5104) + 0.88889 f(2.8280) + 0.55556 f(1.1456)] \\ &\approx -2.1720 [0.55556(1.4616 \times 10^{-9}) + 0.88889(3.3627 \times 10^{-4}) + 0.55556(0.26919)] \\ &\approx -0.32547 \end{aligned}$$

since

$$\begin{aligned} f(4.5104) &= e^{-4.5104^2} \\ &= 1.4616 \times 10^{-9} \end{aligned}$$

$$\begin{aligned} f(2.8280) &= e^{-2.8280^2} \\ &= 3.3627 \times 10^{-4} \end{aligned}$$

$$\begin{aligned} f(1.1456) &= e^{-1.1456^2} \\ &= 0.26917 \end{aligned}$$

b) The absolute relative true error, $|\epsilon_t|$, is (Exact value = -0.31333)

$$\begin{aligned} |\epsilon_i| &= \left| \frac{-0.31333 - (-0.32547)}{-0.31333} \right| \times 100\% \\ &= 3.8757\% \end{aligned}$$