

# Multiple-Choice Test

## Chapter 03.03 Bisection Method

1. The bisection method of finding roots of nonlinear equations falls under the category of a (an) \_\_\_\_\_ method.
  - (A) open
  - (B) bracketing
  - (C) random
  - (D) graphical
2. If  $f(x)$  is a real continuous function in  $[a,b]$ , and  $f(a)f(b) < 0$ , then for  $f(x) = 0$ , there is (are) \_\_\_\_\_ in the domain  $[a,b]$ .
  - (A) one root
  - (B) an undeterminable number of roots
  - (C) no root
  - (D) at least one root
3. Assuming an initial bracket of  $[1,5]$ , the second (at the end of 2 iterations) iterative value of the root of  $te^{-t} - 0.3 = 0$  using the bisection method is
  - (A) 0
  - (B) 1.5
  - (C) 2
  - (D) 3
4. To find the root of  $f(x) = 0$ , a scientist is using the bisection method. At the beginning of an iteration, the lower and upper guesses of the root are  $x_l$  and  $x_u$ . At the end of the iteration, the absolute relative approximate error in the estimated value of the root would be
  - (A)  $\left| \frac{x_u}{x_u + x_\ell} \right|$
  - (B)  $\left| \frac{x_\ell}{x_u + x_\ell} \right|$
  - (C)  $\left| \frac{x_u - x_\ell}{x_u + x_\ell} \right|$
  - (D)  $\left| \frac{x_u + x_\ell}{x_u - x_\ell} \right|$

5. For an equation like  $x^2 = 0$ , a root exists at  $x = 0$ . The bisection method cannot be adopted to solve this equation in spite of the root existing at  $x = 0$  because the function  $f(x) = x^2$
- (A) is a polynomial
  - (B) has repeated roots at  $x = 0$
  - (C) is always non-negative
  - (D) has a slope equal to zero at  $x = 0$

6. The ideal gas law is given by

$$pv = RT$$

where  $p$  is the pressure,  $v$  is the specific volume,  $R$  is the universal gas constant, and  $T$  is the absolute temperature. This equation is only accurate for a limited range of pressure and temperature. Vander Waals came up with an equation that was accurate for larger ranges of pressure and temperature given by

$$\left( p + \frac{a}{v^2} \right)(v - b) = RT$$

Where  $a$  and  $b$  are empirical constants dependent on a particular gas. Given the value of  $R = 0.08$ ,  $a = 3.592$ ,  $b = 0.04267$ ,  $p = 10$  and  $T = 300$  (assume all units are consistent), one is going to find the specific volume,  $v$ , for the above values. Without finding the solution from the Vander Waals equation, what would be a good initial guess for  $v$ ?

- (A) 0
- (B) 1.2
- (C) 2.4
- (D) 3.6

For a complete solution, refer to the links at the end of the book.