

Chapter 08.04

Runge-Kutta 4th Order Method for Ordinary Differential Equations-More Examples

Mechanical Engineering

Example 1

A solid steel shaft at room temperature of 27 °C is needed to be contracted so that it can be shrunk-fit into a hollow hub. It is placed in a refrigerated chamber that is maintained at -33 °C. The rate of change of temperature of the solid shaft θ is given by

$$\frac{d\theta}{dt} = -5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} \theta^4 + 2.33 \times 10^{-5} \theta^3 + 1.35 \times 10^{-3} \theta^2 + 5.42 \times 10^{-2} \theta + 5.588 \right) (\theta + 33)$$

$$\theta(0) = 27 \text{ °C}$$

Using the Runge-Kutta 4th order method, find the temperature of the steel shaft after 86400 seconds. Take a step size of $h = 43200$ seconds.

Solution

$$\frac{d\theta}{dt} = -5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} \theta^4 + 2.33 \times 10^{-5} \theta^3 + 1.35 \times 10^{-3} \theta^2 + 5.42 \times 10^{-2} \theta + 5.588 \right) (\theta + 33)$$

$$f(t, \theta) = -5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} \theta^4 + 2.33 \times 10^{-5} \theta^3 + 1.35 \times 10^{-3} \theta^2 + 5.42 \times 10^{-2} \theta + 5.588 \right) (\theta + 33)$$

$$\theta_{i+1} = \theta_i + \frac{1}{6} (k_1 + 2k_2 + 2k_3 + k_4)h$$

For $i = 0$, $t_0 = 0$, $\theta_0 = 27$

$$k_1 = f(t_0, \theta_0)$$

$$= f(0, 27)$$

$$= \left(-5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} (27)^4 + 2.33 \times 10^{-5} (27)^3 + 1.35 \times 10^{-3} (27)^2 + 5.42 \times 10^{-2} (27) + 5.588 \right) (27 + 33) \right)$$

$$= -0.0020893$$

$$k_2 = f\left(t_0 + \frac{1}{2}h, \theta_0 + \frac{1}{2}k_1h\right)$$

$$= f\left(0 + \frac{1}{2}43200, 27 + \frac{1}{2}(-0.0020893)43200\right)$$

$$\begin{aligned}
&= f(21600, -18.129) \\
&= \left(-5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} (-18.129)^4 + 2.33 \times 10^{-5} (-18.129)^3 \right. \right. \\
&\quad \left. \left. + 1.35 \times 10^{-3} (-18.129)^2 + 5.42 \times 10^{-2} (-18.129) + 5.588 \right) (-18.129 + 33) \right) \\
&= -0.00035761 \\
k_3 &= f\left(t_0 + \frac{1}{2}h, \theta_0 + \frac{1}{2}k_2h\right) \\
&= f\left(0 + \frac{1}{2}43200, 27 + \frac{1}{2}(-0.00035761)43200\right) \\
&= f(21600, 19.276) \\
&= \left(-5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} (19.276)^4 + 2.33 \times 10^{-5} (19.276)^3 \right. \right. \\
&\quad \left. \left. + 1.35 \times 10^{-3} (19.276)^2 + 5.42 \times 10^{-2} (19.276) + 5.588 \right) (19.276 + 33) \right) \\
&= -0.0018924 \\
k_4 &= f(t_0 + h, \theta_0 + k_3h) \\
&= f(0 + 43200, 27 + (-0.0018924)43200) \\
&= f(43200, -54.751) \\
&= \left(-5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} (-54.751)^4 + 2.33 \times 10^{-5} (-54.751)^3 \right. \right. \\
&\quad \left. \left. + 1.35 \times 10^{-3} (-54.751)^2 + 5.42 \times 10^{-2} (-54.751) + 5.588 \right) (-54.751 + 33) \right) \\
&= -0.0035147 \\
\theta_1 &= \theta_0 + \frac{1}{6}(k_1 + 2k_2 + 2k_3 + k_4)h \\
&= 27 + \frac{1}{6}(-0.0020893 + 2(-0.00035761) + 2(-0.0018924) + (-0.0035147))43200 \\
&= 27 + \frac{1}{6}(-0.010104)43200 \\
&= -45.749^\circ\text{C}
\end{aligned}$$

θ_1 is the approximate temperature at

$$t = t_1 = t_0 + h = 0 + 43200 = 43200 \text{ s}$$

$$\theta(43200) \approx \theta_1 = -45.749^\circ\text{C}$$

For $i = 1, t_1 = 43200, \theta_1 = -45.749$

$$\begin{aligned}
k_1 &= f(t_1, \theta_1) \\
&= f(43200, -45.749) \\
&= \left(-5.33 \times 10^{-6} \left(-3.69 \times 10^{-6} (-45.749)^4 + 2.33 \times 10^{-5} (-45.749)^3 \right. \right. \\
&\quad \left. \left. + 1.35 \times 10^{-3} (-45.749)^2 + 5.42 \times 10^{-2} (-45.749) + 5.588 \right) (-45.749 + 33) \right) \\
&= -0.00084673
\end{aligned}$$

$$\begin{aligned}
k_2 &= f\left(t_1 + \frac{1}{2}h, \theta_1 + \frac{1}{2}k_1h\right) \\
&= f\left(43200 + \frac{1}{2}43200, -45.749 + \frac{1}{2}(-0.00084673)43200\right) \\
&= f(64800, -64.038) \\
&= \left(-5.33 \times 10^{-6} \left(\begin{aligned} &-3.69 \times 10^{-6}(-64.038)^4 + 2.33 \times 10^{-5}(-64.038)^3 \\ &+ 1.35 \times 10^{-3}(-64.038)^2 + 5.42 \times 10^{-2}(-64.038) + 5.588 \end{aligned} \right) (-64.038 + 33) \right) \\
&= -0.010012
\end{aligned}$$

$$\begin{aligned}
k_3 &= f\left(t_1 + \frac{1}{2}h, \theta_1 + \frac{1}{2}k_2h\right) \\
&= f\left(43200 + \frac{1}{2}43200, -45.749 + \frac{1}{2}(-0.010012)43200\right) \\
&= f(64800, -262.01) \\
&= \left(-5.33 \times 10^{-6} \left(\begin{aligned} &-3.69 \times 10^{-6}(-262.01)^4 + 2.33 \times 10^{-5}(-262.01)^3 \\ &+ 1.35 \times 10^{-3}(-262.01)^2 + 5.42 \times 10^{-2}(-262.01) + 5.588 \end{aligned} \right) (-262.01 + 33) \right) \\
&= -21.636
\end{aligned}$$

$$\begin{aligned}
k_4 &= f(t_1 + h, \theta_1 + k_3h) \\
&= f(43200 + 43200, -45.749 + (-21.636)43200) \\
&= f(86400, -9.3474 \times 10^5) \\
&= \left(-5.33 \times 10^{-6} \left(\begin{aligned} &-3.69 \times 10^{-6}(-9.4374 \times 10^5)^4 + 2.33 \times 10^{-5}(-9.4374 \times 10^5)^3 \\ &+ 1.35 \times 10^{-3}(-9.4374 \times 10^5)^2 \\ &+ 5.42 \times 10^{-2}(-9.4374 \times 10^5) + 5.588 \end{aligned} \right) (-9.4374 \times 10^5 + 33) \right) \\
&= -1.4035 \times 10^{19}
\end{aligned}$$

$$\begin{aligned}
\theta_2 &= \theta_1 + \frac{1}{6}(k_1 + 2k_2 + 2k_3 + k_4)h \\
&= -45.749 + \frac{1}{6} \left(\begin{aligned} &-0.00084673 + 2(-0.010012) \\ &+ 2(-21.636) + (-1.4035 \times 10^{19}) \end{aligned} \right) 43200
\end{aligned}$$

$$= -45.749 + \frac{1}{6}(-1.4035 \times 10^{19})43200$$

$$= -1.0105 \times 10^{23} \text{ } ^\circ\text{C}$$

θ_2 is the approximate temperature at

$$t = t_2 = t_1 + h = 43200 + 43200 = 86400 \text{ s}$$

$$\theta(86400) \approx \theta_2 = -1.0105 \times 10^{23} \text{ } ^\circ\text{C}$$

The solution to this nonlinear equation at $t = 86400 \text{ s}$ is

$$\theta(86400) = -26.099 \text{ } ^\circ\text{C}$$

Figure 1 compares the exact solution with the numerical solution using Runge-Kutta 4th order method using different step sizes.

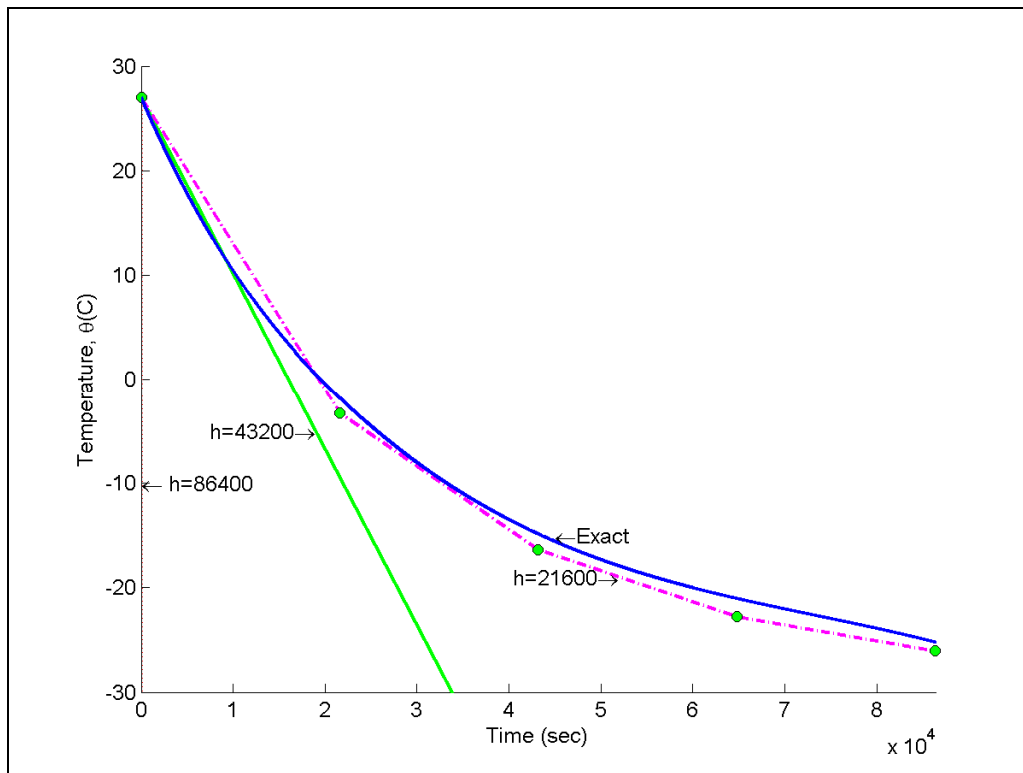


Figure 1 Comparison of Runge-Kutta 4th order method with exact solution for different step sizes.

Table 1 and Figure 2 shows the effect of step size on the value of the calculated temperature at $t = 86400 \text{ s}$.

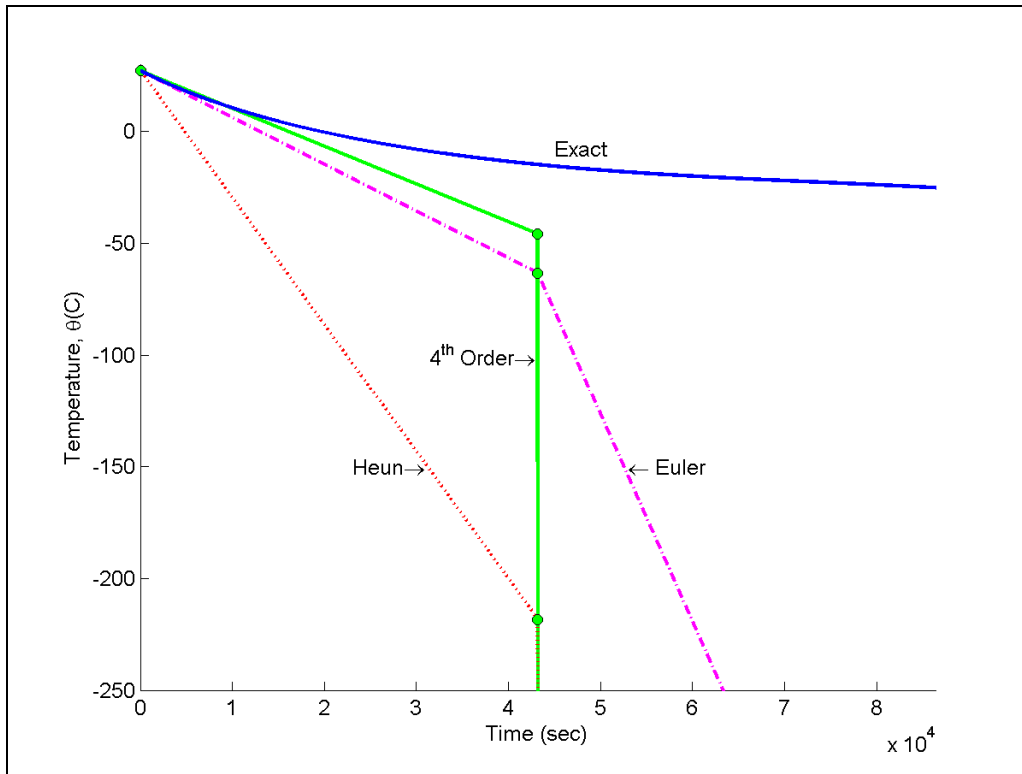


Figure 3 Comparison of Runge-Kutta methods of 1st, 2nd, and 4th order.